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**Research Article** 1 **In-Line Measurement of Color and Total Phenolics** 2 during Red Wine Fermentations Using a 3 **Light-Emitting Diode Sensor** 4 Nicholas L. Shrake, Rajeevan Amirtharajah, Charles Brenneman, 5 Roger Boulton,<sup>2</sup>\* and André Knoesen<sup>1</sup> 6 <sup>1</sup>Department of Electrical and Computer Engineering, University of California, Davis, CA; and 7 8 <sup>2</sup>Department of Viticulture and Enology, University of California, Davis, CA. 9 \*Corresponding author (rbboulton@ucdavis.edu) 10 Acknowledgments: Nicholas L. Shrake is supported by a T.J. Rodgers Fellowship in Electrical and Computer Engineering. The authors gratefully acknowledge the assistance of Babak Taheri and Mark 11 Holst during the project and the donation of the Pinot noir grapes from T.J. Rodgers. 12 Manuscript submitted Feb 2014, revised Jul 2014, accepted Aug 2014 13 14 Copyright © 2014 by the American Society for Enology and Viticulture. All rights reserved. 15 **Abstract:** An in-line color and total phenolic sensor is described to track the color and total phenol 16 17 evolution during red wine fermentations. The sensor uses multiple light emitting diodes (LEDs) spanning the ultraviolet and visible spectrum, in particular 280 and 525 nm. The performance of phenolic sensor 18 19 was evaluated by analyzing fermentation samples collected from multiple red wine fermentations performed during the 2012 season. The study confirmed the LED phenolic sensor measurements strongly 20 correlate with measurements performed with a reference UV-Vis spectrophotometer and that inline 21 22 measurements can be made in a practical manner after removal of yeast and pulp with a 2.0 micron filter. 23 The use of a 100 micron pathlength flow cell avoids the need for dilution, making in-line measurements possible. The sensor provides inline measurement on the evolution of the total phenolic content and color 24 extraction patterns during red wine fermentations. 25 26 27 **Key words:** in-line, real-time, total phenolics, red wine color, fermentation, extraction pattern 28 29

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30 Introduction

In red wine fermentations, color and phenolic compounds are extracted from the grape skins and additional phenolic components from the seeds into the must. The pigments and phenolic compounds extracted during contacting determine the color, and mouth-feel of the wine (Thorngate and Noble 1995, Brossaud et al. 2001). In sensory terms, phenolic compounds are closely associated with astringency and sometimes bitterness and it is of interest to measure these compounds in real-time and to follow the extraction pattern in all stages of contacting and fermentation. Currently, the measurement of phenolics during fermentation involves manually sampling on a regular schedule and storing in a refrigerator, and at a later stage completing the analysis. This approach is slow and the information is delayed and less useful for following and making skin contact decisions in an ongoing fermentation. An inline sensor that monitors phenolic information in real-time is the preferable alternative.

Winemakers make phenolic measurements based on either chromatography or spectroscopy. High performance liquid chromatography (HPLC) allows precise measurement of individual phenolic compounds (Donovan et al. 1998, Peng et al. 2002). While accurate, HPLC systems are technically not feasible for inline setups due to long analysis times (70+ minutes) and significant instrument cost. Spectroscopy measurements can be used to quantify tannin, anthocyanin, and other non-tannin phenols. Color information from the anthocyanins is contained in an absorption band in the 495 - 545 nm region. The total phenol measurement are made around 280 nm, which corresponds to an intense absorption band associated with the benzene ring common to all phenolic compounds. In the simplest optical absorbance analysis, fermentation samples and red wines are analyzed for color at 525 nm and total phenolics at 280 nm (Somers 1998, Harbertson et al. 2006). More specific phenolic compounds can be quantified with chemical assays or multivariate chemometric models for those assays. Chemical assays obtain information by a reaction between the sample and a reagent to produce high absorbance at an indicator wavelength (Harbertson et al. 2003). The use of chemical assays is avoided here however because the procedure is destructive and not adaptable to an inline approach due to the multi-step procedure involving

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long incubation and centrifugation times. Multivariate chemometrics based on NIR and UV-Vis spectra have been explored (Cozzolino et al. 2004, Skogerson et al. 2007) as a tool for rapid and real-time phenolic estimation of subgroups of the total phenols. Of particular importance in red wine fermentation are measurements in the ultraviolet region (250 - 320 nm) where the absorbance is very large (up to 75 AU in a 1 cm path length). The practice of dilution (Skogerson et al. 2007) to bring the UV absorbance on-scale undermines intact color estimation due to the disruption of copigmentation effects (Boulton 2001). Optical measurements can be made on undiluted samples only if a short optical path length, less than 1 mm, is used.

An inline optical spectrophotmeter with a small pathlength cell, 100 to 500 micron, should be able to follow the evolution of color and total phenolics during a red wine fermentation, using undiluted samples. In practice, inline optical measurements are challenging because of light scattering caused by bubbles and suspended particles. While a commercial spectrophotometer could be adapted to perform inline measurements, it is not a desired option because of high cost, power and space involved. A conventional white light source is limited because of low photon flux at most wavelengths, which necessitates expensive photodetectors such a photomultiplier tubes or avalanche-based photodiodes. In contrast, solid-state light emitting diodes (LEDs) offer a low cost option with a large photon flux at the chosen wavelength that allows the use of silicon photodiodes. Custom LED-based photometers have been investigated previously for transmission and reflectance measurement applications in analytical chemistry (Dasgupta et al. 2003). Several groups have adopted LED colorimeters for field trials in processing monitoring (Hauser et al. 1995) and wastewater monitoring (Beaton et al. 2011) because of their low cost, small form factor, and minimal power consumption. The wide absorbance band of the total phenolic and color compounds in red wine made using an LED as an optical source an attractive option.

The aim of this study was to develop and evaluate an inline sensor capable of following the extraction of color and total phenols during red wine fermentations.

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#### **Materials and Methods**

# 1. Inline color and total phenol sensor

The multispectral LED colorimeter is shown in Figure 1. The main factor that determined colorimeter design was the requirement of making absorption measurements in the UV (280 nm and 320 nm) in a quartz transmission flow cell with a short 100 micron path length and a window size of 8 mm x 40 mm (584.4-Q-0.1, Starna Cells; Atascadero, CA). A transmission optical measurement was selected since spectroscopic absorption measurements are performed over a well-defined optical path length. The UV 280 nm LED (UVTOP-280TO18BL; SETI Inc., Columbia, SC) and UV 320 nm LED (UVTOP-325TO18BL; SETI Inc., Columbia, SC) are mounted directly to the flow cell to minimize extraneous absorption of UV photons. Both UV LEDs include a quartz ball lens (spot size of 1.5 mm at a focal point of 2 cm) that is used to improve collection efficiency for UV photons onto the 2.54 mm diameter UV photodiodes (model PC10-2-TO5; Pacific Silicon, Westlake Village, CA). Measurements were also made with six visible wavelength LEDs, 420 nm (OUE8A420Y1; Optek Technology Inc., Carrollton, Texas), 470 nm (WP7113QBC/D; Kingbright, Taipei, Taiwan), 505 nm (OVLGC0C6B9; Optek Technology Inc., Carrollton, Texas), 525 nm (C503B-GAS-CB0F0792; Cree Inc., Durham NC), 565 nm (TLHG4900; Vishay, Malvern, PA), and 630 nm (WP813SRC/F; Kingbright, Taipei, Taiwan). While visible measurements could be made with a longer path length cell, for simplicity of design, it was decided to use the same path length for all measurements. The visible LEDs were coupled into the cell by multimode fluorinated polymer optical fibers (EKSA – (model 02-536; Edmund Optics, Barrington, NJ). The use of polymer fiber to deliver light into the cell was required because of space requirements imposed by the optical flow cell. The visible LED sources were coupled into optical fibers by end coupling. The efficiency of coupling was maximized by grinding the epoxy casing of LEDs to less than 0.5 mm distance from the semiconductor die surface and then optical polishing the surfaces (1-micron grit, model LFG1P; Thor Labs, Newton, NJ). The polished LEDs were mounted on a printed circuit board. The multimode polymer optical fibers are aligned to the LEDs by 15 mm long tubes made into aligning fixture into which

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the fibers are permanently affixed by hot glue (model 2013548; Ace Hardware Brand., Oak Brook, Illinois). The visible wavelength photodiodes (model SFH2701; OSRAM Opto Semiconductor Inc., Munich, Germany) were selected for their small area (1.65 mm x 3.35 mm) and are required due to the small window area of the optical flow cell.

The distance from the UV LED or optical fiber to the photodiode is fixed at 2.6 mm. The numerical aperture for the ball-lens UV LED is 0.05 (3-degree half angle). The numerical aperture for the polymer optical fiber is 0.51 (30-degree half angle). The numerical aperture for the blue-enhanced UV photodiode is 0.53 (32-degree half angle). The numerical aperture of visible-wavelength photodiode is 0.87 (60-degree half angle). The focused beam of the UV LED source fills the UV photodiode surface resulting in maximum collection efficiency. The fiber, owing to its large numerical aperture is a dispersive beam and much of the light falls outside the detectors small area. However, the power coupled into the fiber is sufficiently high that loss is not of concern.

A transimpedance (OPA124; Texas Instruments, Dallas, TX) electronic front-end was used for all visible wavelengths, and a switched integrator (ACF2101; Texas Instruments, Dallas, TX) for the UV wavelengths 280 and 320 nm. The switched integrator timing signals for hold, select, and reset are generated by a microcontroller (PSoC CY8C3866AXI-040; Cypress Semiconductor, San Jose, CA). A bipolar 18-volt power supply (model E3620A; Agilent, Santa Clara, CA) is used with two linear regulators to create 15 Volt and -15 Volt power supplies (MC79M15CDTRKG, MC78M15CDTRKG; ON Semiconductor, Phoenix, AZ). The 15 Volt and -15 Volt supplies are used to power the six transimpedance amplifiers and single switched integrator. The amplifier provides sufficient gain such that the maximum photocurrent will output the full scale system voltage. The maximum photocurrent is around 40  $\mu$ A for visible LEDs and 2  $\mu$ A for the UV LEDs. The gain of all transimpedance channels is set to to 400,000 (V/I) to produce a full scale output of -15 Volts for the 40  $\mu$ A photocurrent. The integration time is set to 500 microsecond for an effective gain of 5,000,000 (V/I) to produce a full scale output voltage of 10 volts for the 2  $\mu$ A photocurrent. The output voltage is level-shifted to a 5 Volt full scale

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before being digitized. The 8 LEDs are sequentially pulsed in a 1 second interval. The transimpedance and integrator outputs are multiplexed and digitized by a 12-bit successive approximation analog to digital converter onboard the microcontroller. The instrument is controlled and data collected over a USB interface by a graphical user interface (LabView, National Instruments, Austin TX). The LED colorimeter makes 64 absorbance measurements at each wavelength and reports the mean and standard deviation of the analysis.

## 2. Red Wine Fermentations:

During the 2012 harvest, 256 samples were collected at various stages during red wine fermentations performed at the UC Davis Department of Viticulture and Enology Pilot Winery. Grapes of the cultivars *Vitis vinifera var*. Syrah, Cabernet Sauvignon, and Barbera fruit were picked from the UC Davis Vineyard and the Cabernet Franc and Malbec was from UC Oakville Vineyard. The fruit was de-stemmed and crushed (Bucher Vaslin, Charlonne-sur-Loire, France) and placed directly into two hundred liter stainless steel research fermentors (Sharpsville Container Corp. Sharpsville, PA). The fermentations were controlled for temperature and mixing was monitored by an integrated fermentation control system developed by Cypress Semiconductor (San Jose, CA). These units were programmed to control a chosen fermentation temperature and were set to automatically pump over the fermentation every three hours. In all, 256 samples were taken from thirteen individual fermentations that included Syrah (156), Cabernet Sauvignon (66), Cabernet Franc (12), Barbera (13), and Malbec (9).

## 3. Inline Study Sample Collection

The evolution of total phenolics and color were followed during the fermentation of two Cabernet fruit and one Pinot Noir fermentations by making inline measurements with the LED sensor. The inline measurement method was validated on two Cabernet Sauvignon fermentations. The block diagram for the fluidic system is shown in Figure 2. Synchronized with fermentation pump overs, the inline system automatically sampled every 5 hours over the fermentation period of 140 hours. A Y-fluid coupler was used to alternate pushing water or wine through the optical cell. A peristaltic pump (model 14-375-40;

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Cole-Parmer, Chicago IL) pulled a wine sample, and pushed the sample through a filter (2.0-micron pore size membrane filter, model AP2504700; Millipore, Billerica, MA, and 47mm in-line polycarbonate filter holder, model 1119; Pall Gelman, Port Washington, NY) and into the flow cell. An identical peristaltic pump pushed deionized water through the flow cell. The inline sampling protocol was as follows: 1.5 minutes of flushing the optical cell with water, 1.5 minutes with water "reference" measurement, 1.5 minutes of wine flowing through the cell, and 1.5 minutes of sample measurement. A total of 100 mL of wine is shunted during the 3-minute pump time. The water and wine flow is halted during the absorbance measurements to avoid bubble contamination. The peristaltic pumps were controlled using LabView (National Instruments, Austin TX) via a solid-state AC relay (Power Switch Tail 2; Adafruit, New York, NY) and a digital interface (model 6009; National Instruments, Austin TX). Control samples were also manually drawn coinciding with the automatic sample times, placed in 50 mL plastic tubes (model 14-375-150; Fisher Brand, Waltham MA) and immediately stored in a refrigerator at -4 degrees C for later analysis.

## 4. Manual Study Sample Collection

Manual samples were collected from 11 fermentations every 12 hours, placed in 50 mL centrifugation tubes (model 14-375-150; Fisher Brand, Waltham MA) and immediately stored in a refrigerator at -4 degrees C. A total of 256 samples were collected and analyzed.

#### 5. Laboratory Sample Analysis

14 mL samples were extracted and transferred to 15 mL plastic centrifuge tubes (model 339650; ThermoFisher, Waltham MA). The extraction and transfer procedure was repeated for the entire manual sample bank (n = 256). The samples were centrifuged (model 225; Fisher Scientific, Waltham, MA) for 10 min and left on a bench for 30 minutes to reach room temperature. UV-visible spectra were collected with a spectrophotometer (model HP 8453; Agilent, Santa Clara, CA) using a 100 µm path length cuvette (584.4-Q-0.1, Starna Cells; Atascadero, CA). Readings were also collected using the LED sensor. The UV-Vis and LED sensor measurements were run on single samples (no replicates). Deionized water was

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used for the reference scan. For both instruments, samples were pumped into the cell using a sipper peristaltic pump (model 89052B; Agilent, Santa Clara, CA) with a 30 second deposit and 10 second wait time before analysis.

183 Results

The performance of the LED colorimeter was validated using an inline and laboratory experimental configuration. The inline study examines the effect of interference from residual particles in the wine solution after a 2.0 micron filtration. Samples were manually drawn at the inline sampling time and centrifuged offline before being analyzed using the LED colorimeter. The study quantifies the absorbance offset contribution from particles by comparing measurements made on red wine fermentation samples using either the 2.0-micron filter or centrifuge. A second investigation was used to determine if the LED colorimeter could be used as a replacement for a UV-Vis spectrophotometer. A large number of fermentation samples were collected and measured using the LED Colorimeter and UV-Vis after a centrifuge step. A strong correlation was found between absorbance measurements made with the LED Sensor and UV-Vis spectrophotometer for total phenolic (R<sup>2</sup>=0.98) and color (R<sup>2</sup>=0.95).

A phenolic extraction plot comparing inline and laboratory absorbance measurements during fermentation is shown in Figure 3. The phenolic measurements shown are made with the LED Colorimeter instrument. The inline method used a pump to push fermentation samples through a filter after which total phenols and red color were measured and recorded. The laboratory method used a centrifuge on samples that were manually collected at the inline sampling times. The absorbance measurements for the inline method were always greater than the laboratory centrifuge system. The absorbance offset is most pronounced for the total phenolic (280 nm) measurement with the background offset of the inline method approaching 50% of the total absorbance measured. In contrast, the color (525 nm) measurement had a background offset of at most 25% of the total absorbance. The background offset arises from suspended particles (yeast and pulp) in the sample. Light is scattered by these particles leading to an effective background absorbance. The inline absorbance offset limits the dynamic range of

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the measurement and obscures the true absorbance value. Standard analytical sample preparation (centrifugation or membrane filtration) removes essentially all particles above a dimension of 0.45-micron. The 0.45 micron membrane filters clog frequently while the 2.0-micron filtration reduces the particle level to a tolerable level in the background absorbance.

The measurement focus of this study was to validate if the LED sensor is a suitable replacement for a UV-Vis spectrometer in making these absorbance measurements. A plot comparing absorbance measured using the LED sensor and UV-Vis for a single Shiraz fermentation is shown in Figure 4. A total of 28 samples were collected during fermentation of which 2 were excluded from this analysis because of interference from carbon dioxide bubbles. The absorbance measured using the LED sensor for the wavelength 280 nm is smaller than the UV-Vis measurement for all samples. The opposite trend was found for the 420 nm LED with the UV-Vis measurement now smaller. The color (525 nm) is approximately equal to the UV-Vis measurement. These results demonstrate that absorbance measurements made with LED sources are closely related to spectral region of the measurement. The broad spectral bandwidth of LED sources results in more or less light in the measurement compared to a narrow bandwidth UV-Vis measurement. Several groups (Macka et al. 1996, Smith and Cantrell 2007) developed a model to relate the effect of broad bandwidth LED sources on absorbance measurements and used it to study a variety of chemical analytes. Smith and Cantrell found that the best linearity is achieved with narrow LED sources centered on broad spectral absorbance features. To test if the relationship between LED and UV-Vis absorbance was systematic, a correlation study was used on all fermentation samples collection during the Fall 2012 season (n = 254). Figure 5 shows a correlation plot comparing LED and UV-Vis absorbance measurements for wavelengths 280, 420, and 525 nm. The absorbance measurements made with LEDs showed high correlation ( $R^2 > 0.94$ ) for both total phenols (280 nm) and red color (525 nm) but performed only adequately ( $R^2 < 0.75$ ) for the 420 nm and other wavelengths. A plot showing phenolic extraction from UV-Vis absorbance measurements of Day 1, 3, and 6 of a red wine fermentation is shown in Figure 6. The slope and shape of the absorbance feature changes very little for

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the total phenols (280 nm) and red color (525 nm) measures with regard to time of extraction. In contrast, the 420 nm and 630 nm shape of the absorbance feature changes substantially over the six fermentation days shown. The poor correlation observed is the result of making a measurement over this non-static absorbance region with a broad spectral bandwidth LED. This is consistent with previous reports (Macka et al. 1996, Smith and Cantrell, 2007) and indicates that LED absorbance measurements are dependent on the overlap of LED and absorbance spectrums. The total phenols and red color absorbance measurements made using the LED sensor were reduced by 40% and 10% respectively compared to the UV-Vis measurement. The high correlations between the two methods for the total phenols and red color allow simple corrections to be applied translating the absorbance into an equivalent UV-Vis measurement.

The phenols extracted during red wine fermentation are strongly related to the fermentation temperature. Three manually sampled fermentations of the same Shiraz fruit were fermented at temperatures of 15, 20, and 25 ° C. The absorbance measurements for total phenols (280 nm) and red color (525 nm) are shown in Figure 7 for a sixteen day fermentation. The rate of total phenol and red color extraction is greater when the fermentation temperature is larger. The absorbance approaches a steady state saturation value for both total phenols and red color suggesting that the rate, but not saturation value is temperature dependent.

246 Discussion

The challenges encountered in developing an inline absorbance sensor can be divided between opto-electronic and fluidic delivery issues. The opto-electronic aspect of the design required a miniaturized jig with minimal distance from LED to photodiode so as to optimize the delivery and capture of photons on the detectors surface. Optimizing the photon flux onto the detector surface improves the signal to noise ratio allowing larger optical densities to be measured. One challenge in using UV LEDs as optical sources is optimizing the light collection on the photodiode detector. The optical power of the UV LED sources is very low  $(300-500~\mu\text{W})$  and matters are further complicated by the poor responsitivity of the photodiodes at the UV wavelengths. The collection efficiency of the LED sensor was optimized by

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using large area photodiodes and minimizing the distance between LED and the detectors. The collection efficiency was good with this distance being fixed at 2.6 mm, the thickness of the flow cell. The maximum photocurrent for the 280 nm LED was measured at 2 µA. Silicon photodiodes with UV transparent windows have a low responsivity of 0.13 A/W at a wavelength of 280 nm. Even under the most optimistic circumstances, in which 100% photon coupling is achieved with no reflective losses, the photocurrent would only be 65 µA. The existing design using a ball lens on the UV LED sources captures nearly all of the light available, as the UV LED spot size is much smaller than the UV photodiode diameter. An alternative approach is to forego the focusing ball lens and to instead use a flat window. The approach requires a large area photodiode and a very short path length between source LED and photodiode. A simple calculation shows that a flat window UV LED with 60-degree view angle and UV photodiode with diameter 2.54 mm diameter would require a path length of 0.7 mm to capture all light. The short path length constraint adds complexity requiring a customized optical flow cell. A ball lens is required in the current LED sensor because of the thickness of the flow cell is 2.6 mm. Future investigations might focus on a miniaturized flow cells where the ball lens is removed to reduce overall system cost.

In chemical analysis, a standard sample preparation is to filter samples using 0.45-micron to remove all suspended particles and microorganisms. Using a 0.45-micron filter for inline sample clarification of must is difficult, with frequent clogging as yeast and pulp accumulate on the membrane surface. Offset correction schemes using reference wavelengths to monitor the extent of optical scattering have been adapted for inline process (Merzlyak et al. 2000, Paz et al. 2002). The 630 nm LED used in this sensor did not successfully quantify the optical scattering in these red wine fermentation measurements. As shown in Figure 4, the change in absorbance of 0.75 AU (in a 1 cm path-length) for the 630 nm LED wavelength is not in the instruments noise and represents a real change in the measured absorbance of roughly 1% transmittance. This 1% change in transmittance is likely attributed to the shoulder of the color absorbance. Another option would be the monitor at a reference wavelength further in the NIR region

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(750 - 2500 nm) for an estimate of the optical scattering. In this region absorbance is essentially independent of phenol or color contributions. The use of background correction schemes using reference wavelength absorbance measurements is advocated because it might reduce the inline filtration requirement. This is important because it will ensure filter life during each fermentation. A 2.0 micron pore filter in a 47 mm housing was used to remove gross particulates from the sampled wine, and effectively removed the major portion of this background absorbance contribution.

The problem of carbon dioxide release leading to bubble formation in the flow cell was also addressed by the inline filtration under pressure of the peristaltic pump. Bubbles can obscure the path length and lead to random variation in the measured absorbance. No additional approach was made in the current experimental setup to avoid or remove bubbles. One approach described for HPLC-based capillary system was sufficient backpressure to avoid outgassing and bubble formation (Wang et al. 2000). Another approach might be to adopt an open-cell dip-type reflectance probe where the sample could be agitated to remove bubbles. Bubble formation is most prevalent about 12 to 36 hours after inoculation of the fermentation when the Brix begins dropping most rapidly. To address bubble interference in the inline trials, the measurements were repeated until consistent results were achieved. In repeating the measurement, the wine pump was turned on to flush the cell with a new sample. Sample measurements are judged to be of poor quality when the standard deviation for any LED wavelength is poor (greater than 0.36% variation from the mean). The primary concern in repeating sample measurements is that it requires frequent replacement of the filter due to a build-up of particles on the membrane surface. Reducing the sample volume or the sampling frequency is critical for prolonging the membrane filter for inline measurements. Future studies might investigate minimizing the volume of wine that flows across the filter to ensure its lifetime during fermentation.

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303 Conclusion

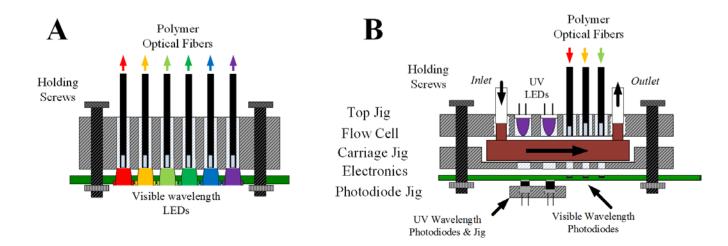
An LED sensor capable of measuring total phenols and red color in real-time during red wine fermentations, is described. The sensor can make absorbance measurements at eight wavelengths: six in the visible, and two in the UV ranges. A short pathlength flow cell was connected to a peristaltic pump to analyze undiluted fermentation samples. The performance of the sensor was tested in two configurations (inline and laboratory) for multiple red wine fermentations. The high correlation found for total phenol (280 nm) and red color (525 nm between the LED sensor and a laboratory spectrophotometer demonstrates the ability to replace a UV-Vis spectrophotometer with the sensor.

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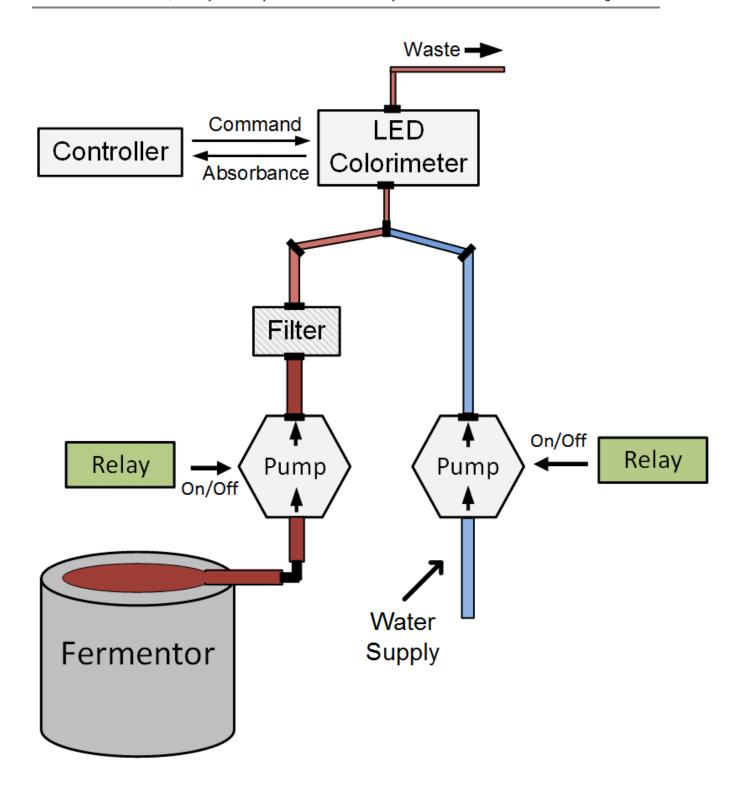
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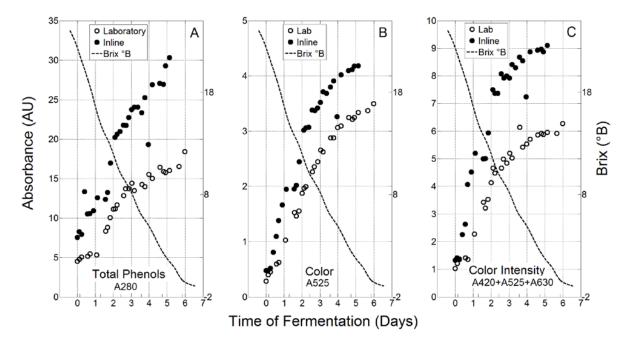
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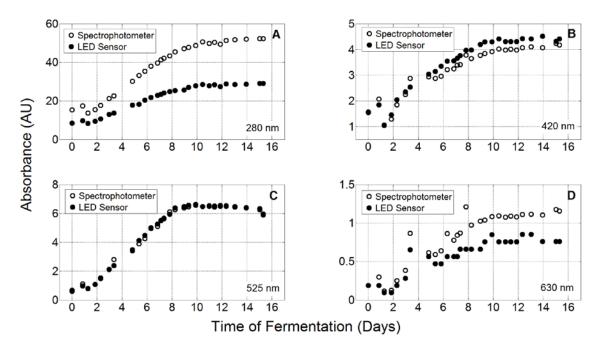
**Figure 1** Cross section of the multi wavelength LED sensor for phenolic and color analysis. (**A**) Optical fibers are polished and coupled to visible wavelength LED sources on a circuit board using a jig. (**B**) Samples are pumped into the inlet of the flow cell. Two UV LEDs are mounted on the cell and 6 visible wavelengths are coupled via polymer optical fibers.



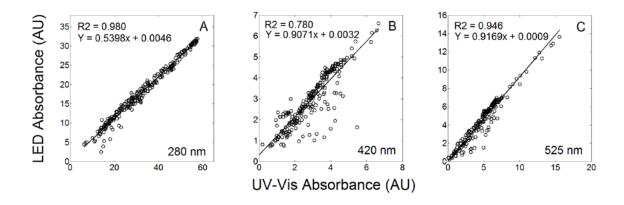
**Figure 2** Block diagram of inline sampling configuration during red wine fermentation. A PC controller applies an ON/OFF signal to a relay block to alternate peristaltic pumps.



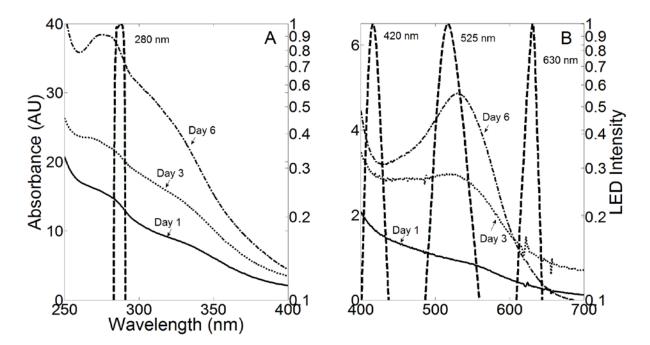
**Figure 3** Extraction curves for, **(A)** Total Phenols, **(B)** Color, and **(C)** Color Intensity for a Cabernet Sauvignon fermentation during an inline experiment. The absorbance was measured using the LED sensor in two configurations: inline (closed circle) and laboratory (open circle) and the Brix curve is overlaid for reference.



**Figure 4** The absorbance measured in a Shiraz fermentation by the LED sensor (open circle) is compared against that measured by a reference UV-Vis spectrophotometer (closed circle). Samples were clarified using a centrifuge before analysis. Absorbance at four wavelengths: (A) 280 nm, (B) 420 nm, (C) 525 nm, and (D) 630 nm.



**Figure 5** The absorbance measured by the LED sensor and UV-Vis spectrophotometer are compared for all fermentation samples (n = 256): Correlation plot for wavelengths (**A**) 280 nm, (**B**) 420 nm, and (**C**) 525 nm.



**Figure 6** The power spectrum of LED light sources is measured and overlaid on some UV-Vis spectra during a red wine fermentation. **A** is for the UV region LED and **B** is for the three visible region LEDs.

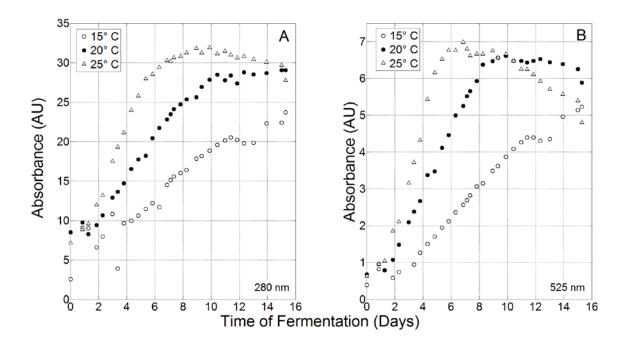


Figure 7 Extraction plot for three Shiraz fermentations with temperature treatments of 15, 20 and 25°C showing (A) total phenols and (B) red color.