# Identifying the Chemical Composition Related to the Distinct Aroma Characteristics of New Zealand Sauvignon blanc Wines

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**Abstract:** A comprehensive set of aroma compounds was quantified in 79 Sauvignon blanc wines from different international producers. Emphasis was given to understanding the chemical differences that can explain the unique character of Marlborough Sauvignon blanc. Quantification revealed the potential importance of several volatile compounds in addition to the polyfunctional mercaptans and methoxypyrazines already known to be important to the aroma of Sauvignon blanc wines. Multivariate statistical approaches, including canonical variate analysis, classification tree and partial least square analysis, established correlations between the chemical and the sensory profiles of the wines. A significant role of 3-mercaptohexyl acetate and 3-mercaptohexanol in the unique tropical, fruity character of Marlborough Sauvignon blanc wines was demonstrated, together with important variations in their concentrations, pointing to different styles even within the Marlborough wines.

Key words: Sauvignon blanc wine, polyfunctional mercaptans, regions, Marlborough

The distinctive sensory characteristics of Sauvignon blanc wines, such as vegetative, grassy, herbaceous, gooseberry, green pepper, boxwood, grapefruit, and passion fruit, have attracted the attention of several flavor chemists. Over the last three decades, many of the compounds responsible for the distinctive Sauvignon blanc aroma have been identified, including two capsicum-like alkyl-methoxypyrazines (2-methoxy-3-isobutylpyrazine [MIBP] and 2-methoxy-3-isopropylpyrazine [MIPP]) (Augustyn et al. 1982, Allen et al. 1991, Lacey et al. 1991, Marais 1994) and three polyfunctional mercaptans (4-mercapto-4-methylpentan-2-one [4MMP], 3-mercapto-hexyl acetate [3MHA], and 3-mercaptohexanol [3MH]), also known as varietal thiols, characterized by boxwood, tropical fruit, and grapefruit characters (Tominaga et al. 1998a, 2006). The hypothesis that these compounds are the principal

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contributors to the aroma of Sauvignon blanc wines is based on their very low perception thresholds resulting in very high odor activity values (OAV: the ratio of concentrations to perception thresholds). However, the wine composition is known to involve numerous additional volatile compounds, and the contribution of fermentative esters, monoterpenes, C<sub>13</sub>-norisoprenoids, and C<sub>6</sub>-alcohols to the complexity of Sauvignon blanc wines has been reported or commented upon in the past (Marais et al. 1994, Francis et al. 1992, Sefton et al. 1994). Moreover, over the last two decades a new dimension has been added to flavor chemistry through the use of aroma reconstitution approaches that allow the sensory interactions of chemical compounds to be examined in progressively more complex matrices (Guth 1997, Lee 2003, Ferreira et al. 2006). These approaches have revealed: (1) the importance of low OAV compounds in the overall wine aroma (Escudero et al. 2004), (2) the aroma enhancement role of  $\beta$ -damascenone (Pineau et al. 2007) and dimethyl sulfide (Escudero et al. 2007), and (3) the suppressing effects of certain compounds, such as the alkyl-methoxypyrazines in lowering the tropical fruit impact of 3MHA (Campo et al. 2005), while ethanol also strongly suppresses wine fruitiness (Escudero et al. 2007). These findings stress the need to expand our consideration of the volatile composition of Sauvignon blanc wines to better understand the overall aroma profile of this variety.

Sauvignon blanc is commercially the most important grape variety in New Zealand and accounts for more than 60% of wine production and for 80% of wine exports. The main region for Sauvignon blanc production in New Zealand is Marlborough, which produces pungent, intense aromatic wines that blend tropical, passion fruit, sweaty, and stone fruit flavors with capsicum and fresh asparagus notes (Lund et al. 2009a). Early studies found higher concentrations of the capsicum-like, herbaceous MIBP in New Zealand Sauvignon blanc

wines (Allen et al. 1991, Lacey et al. 1991), which has been confirmed in more recent studies (Parr et al. 2007, Nicolau et al. 2007, Lund et al. 2009a), where the importance of polyfunctional mercaptans has also been recognized. Higher levels of 3MH and 3MHA in Marlborough wines were established along with good correlations with the tropical passion fruit descriptors, examined using quantitative descriptive analysis (QDA) with a trained sensory panel (Lund et al. 2009a).

The aim of this study was to develop a comprehensive understanding of Sauvignon blanc aroma through the inclusion of more volatile compounds known to contribute to white wine aroma. The emphasis of this study was on the Marlborough region, thus more wines were sampled from this region, together with wines from the Hawkes Bay and Wairarapa regions in New Zealand and from the United States, South Africa, Chile, Australia, and France, all of which produce quality wines of different styles for the international market.

### **Materials and Methods**

Samples. The study was conducted over two years with commercial wines mainly from two vintages, 2004 and 2005 (Table 1). Since the wines were selected from countries of both hemispheres, there were differences in the date of bottling and the age of the wines at analysis. For both sets of wines, the wines from the southern hemisphere from the 2004 and 2005 vintages were analyzed approximately 14 months after bottling. The wines from the northern hemisphere were analyzed approximately 8 months or 20 months after bottling. The first set of northern hemisphere wines were from the 2004 vintage, except for three wines from France and five wines from the United States, which were from the 2003 vintage. The second set of northern hemisphere wines were from the 2005 vintage, except for two French wines from the 2004 vintage. The wine samples were assessed in triplicate for both chemical and sensory analyses. For the first set of samples, 51 wines were analyzed by sensory and chemical analysis. When sensory analysis was performed, five bottles of wines were sampled for each wine; three were used for sensory analysis and two for chemical analysis. In order to eliminate variation due to the timing of the analysis and bottle-to-bottle variation, the two bottles were opened at once and adequate aliquots were taken for each chemical analysis and kept frozen (-20°C) for analysis, which were conducted within four months. The chemical

 Table 1
 Origin and vintage of two sets of commercial wine

 samples from international regions and three New Zealand regions.

•	_			-	
	Firs	t set	Second set		
Country/region	2003	2004	2004	2005	
Australia		5		4	
Chile				3	
France	3	2	2	1	
South Africa		6			
United States	5			3	
Hawkes Bay (NZ)		7		3	
Marlborough (NZ)		16		9	
Wairarapa (NZ)		7		3	

analysis was conducted by category of chemicals, starting with polyfunctional mercaptans, methoxypyrazines, and finishing with the remaining compounds. The triplicate analysis of each sample was conducted on the same day from the same aliquot. The sensory analysis started two months after the beginning of the chemical analyses and was completed within three months.

An additional set of 24 Marlborough commercial Sauvignon blanc wines from the 2007 vintage were quantified for their concentrations of polyfunctional mercaptans approximately eight months after bottling. The retail price of the wines analyzed in this study ranged from US\$6 to \$20 per bottle, with most of the wines priced between \$8 and \$14.

**Reagents.** All of the chemicals used were of analytical quality. Sodium hydroxide, hydrochloric acid, sodium sulfate, sodium acetate, sodium dihydrogen phosphate dihydrate, and diethyl ether were from Scharlau (Sentmenat, Spain). Dowex Resin 1X2 Cl<sup>-</sup>-form, L-cysteine hydrochloride hydrate, ethyl acetate, and *p*-hydroxymercuribenzoate were from Sigma-Aldrich (Steinheim, Germany). Hexane was from Burdick & Jackson (Muskegon, MI), Tris from AppliChem (Darmstadt, Germany), DTNB from Acros Organics (Morris Plains, NJ), ethanol from Ajax Finechem (Taren Point, NSW, Australia), and dichloromethane from Merck (Darmstadt, Germany).

Isoamyl acetate (>98%) was from Ajax Finechem. Ethyl butanoate (99%), hexyl acetate (99%), ethyl octanoate (>99%), ethyl decanoate (>99%), diethyl succinate (99%), 1-hexanol (98%), phenylethanol (>99%), 2-methoxy-3-isobutylpyrazine (97%), 2-methoxy-3-isopropylpyrazine (99%), linalool (97%), citronellol (95%), geraniol (99%), 3-octanol (>99%) and 4-methoxy-2-methyl-2-mercaptobutane (99%) were from Acros Organics. Ethyl hexanoate (>98%), isobutyl acetate (>99.8%), ethyl isovalerate (>99.7%), ethyl isobutyrate (99%), 3-hexenol (Z) (98%), methionol (99%), 2-hexenol (Z) (95%), and nerol (97%) were from Sigma-Aldrich. 3-Hexenol (E) (97%), 2-hexenol (E) (97%),  $\alpha$ -terpineol (97%), and 2-methoxy-3methylpyrazine (98%) were from Lancaster (Ward Mill, MA). Isoamylalcohol (98.5%) was from Panreac Quimica (Barcelona, Spain). Isobutanol (99.5%) was from Scharlau. 3-Mercaptohexanol (98%) and 4-mercapto-4-methylpentan-2-one (1% in polyethylene glycol) were from Interchim (Montluçon, France). 3-Mercaptohexyl acetate (98%) was from Oxford Chemicals (Billingham, UK).

Quantitative analysis. A specific extraction of wine polyfunctional mercaptans was undertaken according to previous methods (Tominaga et al. 1998b, 2006) from 50 mL wine. In the first year of study, the thiols were quantified using 4-methoxy-2-methyl-2-mercapto-butane as internal standard. For varietal thiol quantification in the second year, the deuterated analogs of 3-mercaptohexanol (3-mercapto(1-2H<sub>2</sub>)hexanol) and 3-mercaptohexyl acetate (3-mercapto(1-2H<sub>2</sub>)hexyl acetate) were used as internal standards (Hebditch et al. 2007). Deuterated standards were available only in the second year of the study and helped to improve the method precision (Table 2).

The methoxypyrazines were quantified by liquid-liquid extraction using ether:hexane (1/1, v/v) as described elsewhere (Kotseridis et al. 1999) from 200 mL wine and using

2-methoxy-3-(<sup>2</sup>H<sub>3</sub>)isobutylpyrazine instead of 2-methoxy-3-(<sup>2</sup>H<sub>2</sub>)isobutylpyrazine as the internal standard. All other compounds, including the esters, alcohols, and monoterpenes, were quantified in the same extract prepared for the methoxy-pyrazine quantification, using 3-octanol as the internal standard. When the extract from the methoxypyrazine analysis was concentrated to 50%, an aliquot was taken for quantification of esters and further volatile compounds using the chromatographic conditions described by Sabon et al. (2002).

All samples were analyzed with an Agilent 6890N gas chromatograph (Santa Clara, CA) and a 5973 mass selective detector. The MS was in EI mode, with EMV at 1953 V, the

source at 230°C, the quad at 150°C, emission at 34.6  $\mu$ A, and the Ele energy at 69.9 eV. The capillary columns were BP20 (SGE Analytical Science, Ringwood, Australia) (50 m x 220  $\mu$ m x 0.25  $\mu$ m) for the first set of samples analyzed and HP-INNOWax (Agilent) (60 m x 252  $\mu$ m x 0.25  $\mu$ m) for the second set of samples. All of the compounds were identified based on the retention times of reference compounds. Linear retention indices on the capillary columns agreed with those given in the literature for a C20M column with similar properties (www.flavornet.com).

Calibrations were carried out in all cases using the standard addition method. A wine sample was extracted together

Table 2 Linearity, quantification data, perception thresholds, and descriptors for the volatile compounds analyzed in this study.

Compound	Correlation coefficient (r <sup>2</sup> )	Calibration range (µg/L)	Recovery in calibration range (% )	Average relative standard deviation (%)	Perception threshold (µg/L)	Descriptors
Esters						
Isoamyl acetate <sup>a</sup>	0.998	1194-3980	96-104	4.3	50	Banana, fruity
Hexyl acetate <sup>a</sup>	0.995	52-870	90-107	3.8	400	Apple, cherry, pear, flower
Ethyl butanoate <sup>b</sup>	0.999	96-960	93-109	6.2	20	Fruity, apple
Ethyl hexanoatea	0.996	212-1414	96-113	3.4	45	Apple peel, fruit
Ethyl octanoate <sup>c</sup>	0.997	684-2851	93-104	3.9	600	Sweet, ripe banana, pear
Ethyl decanoated	0.995	92-920	94-116	5.2	200	Fruity, floral
Diethyl succinate <sup>c</sup>	0.999	137-911	98-107	3.8	1250000	Fruit
Isobutyl acetate®	0.982	12-119	91-122	6.8	1605	Fruit, apple, banana
Ethyl isovalerated	0.996	3.6-109	95-112	5.8	3	Fruit
Ethyl isobutyrateb	0.995	10–167	93-106	9.0	15	Sweet, rubber
Alcohols						
1-Hexanol <sup>g</sup>	0.982	342-5693	94-108	3.6	1110	Resin, flower, green, cut grass
3-Hexenol ( <i>E</i> )°	0.993	11-112	91-117	3.9	1000	Green, cut grass
3-Hexenol (Z) <sup>b</sup>	0.993	50-841	87-114	3.5	400	Green, cut grass
Phenylethanold	0.999	13900-46330	98-104	4.8	14000	Honey, spice, rose, lilac
Isoamylalcohol <sup>b</sup>	0.989	29700-148300	83-110	5.2	30000	Whiskey, malt, burnt
Isobutanol <sup>b</sup>	0.987	4000-36030	92-112	8.0	40000	Wine, solvent, bitter
Methionol <sup>b</sup>	0.994	157-2624	90-111	3.6	500	Sweet, potato
2-Hexenol (E)9	0.998	0.55-7.8	83-105	12.6	_	Green, cut grass
2-Hexenol (Z) <sup>g</sup>	0.999	1.3-58.1	90-102	8.2	-	Green, cut grass
Monoterpenes						
Linalool <sup>d,g</sup>	1	5.09-15.9	99-100	5.4	25.2	Fruity, citrus, floral, lavender
Terpineol <sup>g,I</sup>	0.998	2.93-19.1	96-111	3.9	250	Floral, sweet
Citronellol <sup>f,g</sup>	0.999	0.66 - 2.2	99-100	4.6	100	Rose
Nerol <sup>g</sup>	0.999	0.7 - 2.35	_	_	_	Fruity, floral, sweet
Geraniol <sup>b,g</sup>	0.999	0.82-2.75	99–101	5.1	300	Rose, geranium
Thiols						
3-Mercaptohexanol <sup>h</sup>	0.998	0.458-10.98	90-105	5.1 (13.1) <sup>k</sup>	0.06	Grapefruit, passion fruit
3-Mercaptohexyl acetateh	0.999	0.052-1.678	95-106	7.1 (15.3) <sup>k</sup>	0.004	Passion fruit, box tree
4-Mercapto-4-methylpentan-2-one	0.999	0.0054-0.172	97–104	9.4	0.0008	Box tree, broom, cat urine
Methoxypyrazines	0.000	0.0045 0.070	94–104	F	0.000	Croop popper
2-Methoxy-3-isobutylpyrazine <sup>i</sup>	0.998 0.999	0.0045-0.072 0.0046-0.074	94–104 98–108	5.5 4.8	0.002 0.002	Green pepper
2-Methoxy-3-isopropylpyrazine <sup>j</sup>	0.999	0.0040-0.074	90-108	4.0	0.002	Green pepper, earthy

<sup>&</sup>lt;sup>a</sup>12.5% (v/v) ethanol, pH 3.2.

b10% (w/w) ethanol (Guth 1997).

<sup>°14% (</sup>v/v) ethanol, pH 3.5 (Moyano et al. 2002).

d10% (v/v) ethanol, pH 3.2 (Ferreira et al. 2000).

e10% (v/v) ethanol, pH 3.2 (Ferreira et al. 2002).

<sup>&</sup>lt;sup>f</sup>10% (v/v) ethanol, pH 3.5 (Peinado et al. 2006).

<sup>&</sup>lt;sup>9</sup>Quantified only in the first year of the study. h12% (v/v) ethanol (Tominaga et al. 1998a).

White wine (Allen et al. 1991).

Water (Seifert et al. 1970); average relative standard deviation (%).

kAverage of 49 triplicates, first year using 4-methoxy-2-methyl-2-mercaptobutane as internal standard.

<sup>&</sup>lt;sup>1</sup>Media unknown (Rychlik et al. 1998).

with increasing concentrations of the compounds to be quantified. Peak areas for the analytes and internal standards were integrated and the peak areas of the analytes were divided by the peak areas of the corresponding internal standard. The ratio from the blank wine was then subtracted from the ratios of the samples containing increasing amounts of the aroma compounds. Linear regression was used to obtain the function to calculate the concentration of the analytes in the samples. The values (Table 2) were obtained during method evaluations undertaken in the second year of analysis, with the exception of a few compounds that were only quantified in the first study year. R<sup>2</sup> values, recovery, and average relative standard deviations (RSD) were very similar for the analyses in both years, except for 3MH and 3MHA where the quantification precision improved in the second year of quantification, from 13.1 and 15.3 average %RSD, to 5.1 and 7.1 average %RSD, respectively, due to the use of labeled standards.

**Sensory analysis.** Descriptive analysis of the wines was carried out under previously described conditions (Lund et al. 2009a). Trained panelists evaluated 51 wines in triplicate under an incomplete randomized block design. The panelists were given the samples randomly and the randomized samples were blocked by replication (1, 2, 3). The panelists evaluated 10 to 11 wines per session, with a 30-sec break after each wine and a 5-min break after every three wines. Each panelist returned for 15 sessions so that each one tasted every wine. Variations were made to the presentation order of the wine samples served concurrently to all panelists and to the presentation order of subsequent replicate samples provided to individual panelists. The panelists rated the intensities of 16 attributes on a computer using Compusense software (version 5.0, Guelph, Canada). The attributes and their associated reference standards are listed (Table 3).

**Statistical analysis.** Microsoft Excel 2007 software was used for basic data analysis and generation of boxplots. For

the identification of regional differences in the levels of aroma compounds, one-way ANOVA was carried out in SAS 9.2 (SAS Institute, Cary, NC) with all pair-wise comparisons adjusted for multiple comparisons using Tukey's HSD at the 95% confidence level. For a multivariate examination of the data, canonical variate analysis (CVA) was used and a classification tree was fitted in Genstat 10 and used the Ginni Information Criterion to select the variate to split the data at each node. A minimum number of five observations in each group were specified to stop the splitting process. Partial least squares regression (PLS) was performed (Unscrambler version 9.1, CAMO, Oslo, Norway) to determine relationships among the chemical and sensory data.

# Results

Quantifications are given for the first set (2003 and 2004 vintages; Table 4) and second set (2004 and 2005 vintages; Table 5) of wines, summarized as the mean and standard deviation values for each region and country. One-way analysis of variance (ANOVA) was used as a test for the difference in the means of each chemical (*p* value). When the overall *p* value was significant, pair-wise comparisons were adjusted for multiple comparisons using Tukey's HSD at the 95% confidence level to compare the regions with protection against false positives due to multiple testing.

Some additional esters—ethyl hexanoate, ethyl isovalerate, ethyl isobutyrate, diethyl succinate, and isobutyl acetate—were included in the second year (Table 5) following GC–MS and GC/O screening (Nicolau et al. 2008). Most esters quantified in this study had concentration ranges similar to those reported for other white varieties (Guth 1997, Ferreira 1993), except for a few compounds with higher (including ethyl hexanoate and ethyl isovalerate) or lower (such as ethyl isobutyrate) concentrations. As indicated by the *p* values and the pair-wise comparison tests (Table 4), in the first set of

Table 3 Sauvig	non blanc sensory reference standards used in trained panel evaluations (Lund et al. 2009a).				
Lexicon	Reference standard				
Sweet sweaty passion fruit	2,000 ng/L 3-mercaptohexyl acetate (Oxford Chemicals) <sup>a</sup>				
Capsicum	1,000 ng/L 2-methoxy-3-isobutylpyrazine (Acros Organics) <sup>a</sup>				
Cat urine/box wood	1,000 ng/L 4-mercapto-4-methylpentan-2-one (Oxford Chemicals) <sup>a</sup>				
Passion fruit skin/stalk	2,000 ng/L 3-mercaptohexan-1-ol (Interchim) <sup>a</sup>				
Grassy	28,800 ng/L 2-hexenol (Z) (Sigma) <sup>a</sup>				
Flint/mineral	4,000 ng/L benzyl methyl thiol (Oxford Chemicals) <sup>a</sup>				
Citrus	30 g Yen Ben lemon plus 15 g Bear lime soaked in diluted base wine for 30 min <sup>b</sup>				
Bourbon	2,400 μg/L hexanol (Sigma) <sup>a</sup>				
Apple Iolly/candy	2,500 μg/L hexyl acetate (Sigma) <sup>a</sup>				
Tropical	40 mL Golden Circle Mango juice + 40 mL Golden Circle Golden Pash + 200 mL Just Juice Mandarin Passion Fruit juice <sup>b</sup>				
Mint	25 mg/L cineole (Sigma) <sup>a</sup>				
Fresh asparagus	50 mL steamed asparagus water <sup>b</sup>				
Canned asparagus	10 mL Watties canned asparagus juice <sup>b</sup>				
Stone fruit	Canned Watties apricot and peach juice soaked in diluted base wine for 30 min (equal parts) <sup>b</sup>				
Apple	70 g Sciros/Pacific Rose apple peeled, soaked in diluted base wine for 30 min <sup>b</sup>				
Snow pea	1,275 ng/L 2-methoxy-3-methylpyrazine (Acros Organics) <sup>a</sup>				

<sup>&</sup>lt;sup>a</sup>Added to the diluted base wine (50% Corbans Sauvignon blanc and 50% water).

<sup>&</sup>lt;sup>b</sup>Added in equal parts to the base wine (Corbans Sauvignon blanc).

samples most esters were significantly higher in the three New Zealand wine regions than in the international wines, but this difference was not repeated with the second set of samples (Table 5). Significant differences between the regions and countries and for both sets of samples were also observed in the mean levels of the higher alcohols.

The monoterpenes were quantified only in the first set of samples, and the levels agreed with those previously reported for Sauvignon blanc wines (Ribéreau-Gayon 2001). Significant differences were observed between regions and countries, except for terpineol (Table 4).

More evident differences between the regions and countries were observed for the varietal C<sub>6</sub>-alcohols, methoxypyrazines, and polyfunctional mercaptans. Among the C<sub>6</sub>-alcohols, 3-hexenol (*Z*) stood out with significantly higher concentrations in wines from the Marlborough region, followed closely by wines from other New Zealand regions and Australia. 2-Methoxy-3-isobutylpyrazine (MIBP) and 2-methoxy-3-isopropylpyrazine (MIPP) were quantified in the first set of samples (Table 3). The mean values for MIPP were slightly higher than for concentrations previously reported in Sauvignon blanc wines (Lacey et al. 1991). Minor but significant differences were observed between the regions and countries, notably significantly higher mean levels in the Australian wines. As previously demonstrated (Lund et al. 2009a), the New Zealand wines, and especially those from

the Wairarapa and Marlborough, were significantly higher in MIBP than the wines from other countries, in agreement with previous studies (Lacey et al. 1991). Due to an instrumental fault at the time of running the second set of samples, only MIBP was quantified (Table 5). The mean level for the Marlborough wines was very close to that obtained for the previous set of samples, but the mean level for the Wairarapa wines was lower. The missing data for some Hawkes Bay, Australia, and French wines made the MIBP data insufficient for statistical analyses.

The polyfunctional mercaptans 3MH, 3MHA, and 4MMP are important impact odorants in Sauvignon blanc wines, and the quantitative data obtained in this study were consistent with concentrations reported previously (Tominaga et al. 1998a). For 3MHA, the mean concentration for Marlborough wines was significantly higher than that of all other regions and countries for the first set of samples (Table 4). The second set was similar (Table 5), with the exception of the Chilean and United States means, which were not significantly different than the Marlborough values. Likewise, for 3MH with the first set of samples (Table 4), the mean concentration in the Marlborough wines was significantly higher than that of all of the other regions and countries except Wairarapa. However, in the second set of samples, 3MH mean for the Marlborough wines was only significantly different than the Australian, French, and United States mean (Table 5). In the first set of

**Table 4** Average concentrations, standard deviations, and ANOVA *p* values for the volatile compounds quantified in the first set of samples (2003 and 2004 vintages).

	Marlborough <sup>m</sup> (NZ) n = 7	Hawkes Bay <sup>h</sup> (NZ) n = 7	Wairarapa <sup>w</sup> (NZ) n = 7	Australiaª n = 5	South Africas n = 6	France <sup>f</sup> n = 5	USA <sup>u</sup> n = 5	<i>p</i> value
Esters								
Ethyl butanoate (µg/L)	548 ± 99 <sup>f*</sup>	576 ± 100 <sup>f</sup>	510 ± 89	381 ± 81 <sup>u</sup>	$480 \pm 154$	378 ± 122 <sup>9</sup>	$680 \pm 446$	0.0599
Ethyl octanoate (µg/L)	1371 ± 479s,u	$1455 \pm 516^{s,u}$	$1631 \pm 359^{a,s,u}$	1116 ± 211	$864 \pm 367$	1265 ± 562	$790 \pm 98$	0.0092
Ethyl decanoate (µg/L)	368 ± 197 <sup>u</sup>	362 ± 221 <sup>u</sup>	$464 \pm 213^{a,s,f,u}$	271 ± 140	194 ± 228	227 ± 148	78 ± 12	0.0165
Isoamyl acetate (µg/L)	$2531 \pm 670^{a,s,f,u}$	$2515 \pm 963^{a,f,u}$	$2892 \pm 962^{a,s,f,u}$	1430 ± 371	$1732 \pm 773$	$1547 \pm 805$	$999 \pm 345$	0.0001
Hexyl acetate (µg/L)	$196 \pm 71^{a,s,f,u}$	$174 \pm 117^{a,s,f,u}$	$148 \pm 32^{s,f,u}$	$88 \pm 20$	$56 \pm 37$	$43 \pm 27$	$43 \pm 31$	<.0001
Alcohols								
1-Hexanol (µg/L)	3081 ± 617s,f	2901 ± 903s,f	$3105 \pm 536^{s,f}$	2710 ± 602s,f	$1672 \pm 440^{\circ}$	1732 ± 942 <sup>u</sup>	2872 ± 1190	0.0009
3-Hexenol (E) (µg/L)	98 ± 16ª	69 ± 22a	$94 \pm 26^{a}$	$175 \pm 90^{s,f,u}$	$98 \pm 37$	102 ± 85	$95 \pm 65$	0.0235
3-Hexenol (Z) (µg/L)	$599 \pm 179^{w,a,s,f,u}$	$528 \pm 159^{s,f,u}$	$417 \pm 91^{s,f,u}$	$420 \pm 95^{f,u}$	$260 \pm 99$	197 ± 84	$227 \pm 46$	<.0001
2-Hexenol (E) (µg/L)	$2.8 \pm 2.6^{s}$	$2.9 \pm 2.1$	$2.7 \pm 1.3$	$4.5 \pm 4.1$ s,u	$0.5 \pm 0.8^{f}$	$4.4 \pm 3^{u}$	1.1 ± 1.0	0.0329
2-Hexenol (Z) (µg/L)	12 ± 3 <sup>h</sup>	$22 \pm 9.8^{w,a,s,f,u}$	15 ± 3	15 ± 5	10 ± 5	$15 \pm 5$	$15 \pm 6$	0.0022
Phenylethanol (mg/L)	$38 \pm 16^{a,s,u}$	33 ± 11	$43 \pm 14^{a,s,u}$	$22 \pm 10^{f}$	17 ± 4 <sup>f</sup>	$46 \pm 30^{u}$	19 ± 8	0.0038
Monoterpenes								
Linalool (µg/L)	$12 \pm 4.2^{w,s}$	12 ± 3s	$15 \pm 4^{a,s,u}$	$9.8 \pm 3.8$	$7.0 \pm 3.1^{f}$	$13 \pm 6$	$8.8 \pm 2.2$	0.005
Terpineol (µg/L)	19 ± 6	18 ± 3	18 ± 5	21 ± 7	15 ± 5	19 ± 9	14 ± 4	0.3666
Citronellol (µg/L)	$2.6 \pm 0.7^{s,f}$	$2.5 \pm 0.4^{f}$	$2.3 \pm 0.7$	$2.1 \pm 0.6$	$1.8 \pm 0.5^{u}$	$1.7 \pm 0.4^{u}$	$2.6 \pm 0.7$	0.0282
Geraniol (µg/L)	$2.0 \pm 0.8^{s}$	$2.2 \pm 0.7^{s}$	$2.8 \pm 0.9^{a,s,u}$	1.6 ± 1.1	$1.0 \pm 0.4^{f}$	$2.3 \pm 1.7^{u}$	$1.3 \pm 0.5$	0.0061
Thiols								
3MHA (ng/L)	$516 \pm 572^{h,w,a,s,f,u}$	$65 \pm 45$	99 ± 121	$68 \pm 10$	$38 \pm 23$	$29 \pm 34$	$45 \pm 18$	0.0001
3MH (ng/L)	$7080 \pm 5567^{h,a,s,f,u}$	1733 ± 765 <sup>w</sup>	5219 ± 5316s	2379 ± 1664	1526 ± 376	$2050 \pm 869$	2094 ± 1628	0.0006
4MMP (ng/L)	$8.2 \pm 3.4$	$9.0 \pm 5.1$	$19 \pm 9^{a,s,u}$	$5.5 \pm 2.7$	$6.6 \pm 1.5$	$9.7 \pm 4.5$	$5.7 \pm 2.1$	0.003
Methoxypyrazines								
MIBP (ng/L)	$22 \pm 7^{w,a,s,f,u}$	$17 \pm 9.4^{w,f,u}$	$35 \pm 9^{a,s,f,u}$	14 ± 4 <sup>u</sup>	14 ± 11 <sup>u</sup>	$7.9 \pm 4.2$	$4.6 \pm 2.4$	0.003
MIPP (ng/L)	$8.4 \pm 1.8^{h,a}$	$7.4 \pm 0.8^{w,a}$	$9.5 \pm 1.7^{a,s,u}$	$13 \pm 3^{s,f,u}$	$7.9 \pm 1.2$	$8.1 \pm 2.6$	$7.8 \pm 0.7$	<.0001

<sup>\*</sup>Superscript letters (m = Marlborough, h = Hawkes Bay, w = Wairarapa, a = Australia, s = South Africa, f = France, u = United States) indicates the significant differences between countries and regions obtained using a pair-wise comparison (Tukey's HSD) with 95% level of confidence.

samples, the 4MMP mean (Table 4) of the Wairarapa wines was significantly higher than the mean for the Australian, South African, and United States wines. In the second set of samples, despite higher means obtained again for the Wairarapa and United States wines, no significant differences in 4MMP concentrations were observed (Table 5).

The differences among regions and countries in the second set of samples were not as great as in the first set, especially regarding the esters. The lack of significant differences observed with the second set can be explained by the low number of samples quantified, as only three wines were analyzed for most of the countries and regions, except for Australia with four wines and Marlborough with nine wines, which made this set of data not suitable for further statistical investigation. Moreover, in the first set of samples, 3MH and 3MHA concentrations in some of the Marlborough wines were particularly high and were also characterized by a greater variation within the region (Figure 1). However, with the second set of samples, the average 3MH and 3MHA concentrations were lower and the variation was not so extreme. As the 2006 Marlborough vintage had the earliest harvest dates on record, the testing of the additional Marlborough Sauvignon blanc wines from the 2007 vintage was important to confirm that the high 3MH and 3MHA concentrations obtained in the first set of wines were not accidental. The results for 3MH and 3MHA for these extra 2007 wines (Figure 2)

were closer to those obtained for the 17 wines from the first set of wines (2004 vintage) than for the nine wines from the second set (2005 vintage). Moreover, the wide Marlborough within-region variation observed for the 2004 vintage wines was confirmed in this larger sample set. The 4MMP concentrations obtained for the 2007 vintage Marlborough wines ranged from 6.2 to 19.5 ng/L (data not shown).

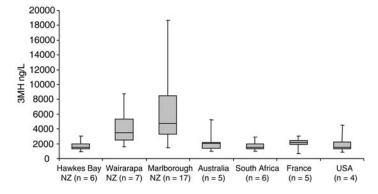
The higher number of samples (51 wines) analyzed for the first data set allowed the application of multivariate statistical analysis for the identification of regional differences. Canonical variate analysis (CVA) was chosen for its ability to maximize differences between groups. A CVA plot was obtained using concentrations of aroma compounds and regions as variables (Figure 3). The ellipses represent the 95% confidence intervals for the means of the regions. Fifty percent of the differences in this data set can be explained by the first two dimensions. The first dimension explains 26% of the difference and separates the three New Zealand regions from the other countries (Figure 3A), which was due to their higher concentrations of 3-hexenol (Z), hexyl acetate, isoamyl acetate, 3MH, 3MHA, 1-hexanol, ethyl octanoate, and ethyl decanoate. In the second dimension, Wairarapa was separated from the other two New Zealand regions by higher levels of 4MMP, MIBP, and the two terpenes linalool and geraniol. The 95% confidence intervals for the Australian and South African wines overlap. Australia, South Africa, and United

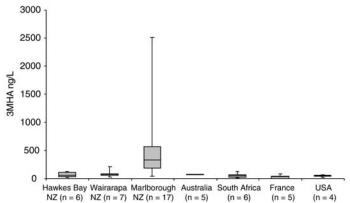
**Table 5** Average concentrations, standard deviations, and ANOVA *p* values for the volatile compounds quantified in second set of samples (2004 and 2005 vintages).

	Marlborough <sup>m</sup> (NZ) n = 9	Hawkes Bay <sup>h</sup> (NZ) n = 3	Wairarapa <sup>w</sup> (NZ) n = 3	Australia <sup>a</sup> n = 4	Chile <sup>c</sup> n = 3	France <sup>f</sup> n = 3	USA <sup>u</sup> n =3	<i>p</i> value
Esters								
Ethyl butanoate (µg/L)	337 ± 102 <sup>w,a,u*</sup>	441 ± 83	$487 \pm 89$	462 ± 132	$451 \pm 69$	$359 \pm 34^{\circ}$	$530 \pm 89$	0.045
Ethyl hexanoate (µg/L)	1168 ± 1491	1137 ± 122	1330 ± 124	1248 ± 242	1250 ± 150	1042 ± 111	1155 ± 214	0.436
Ethyl octanoate (µg/L)	1697 ± 394	1695 ± 333	$2505 \pm 926$	1950 ± 296	1650 ± 263	$1629 \pm 59$	1578 ± 262	0.118
Ethyl decanoate (µg/L)	581 ± 166	$574 \pm 309$	$608 \pm 164$	$497 \pm 292$	$450 \pm 80$	$416 \pm 80$	$435 \pm 27$	0.669
Ethyl isobutyrate (µg/L)	$69 \pm 27$	$96 \pm 41$	$105 \pm 64$	$85 \pm 24$	$113 \pm 30$	97 ± 22	$70 \pm 16$	0.317
Ethyl isovalerate (µg/L)	14 ± 6	21 ± 8	22 ± 10	19 ± 5	$24 \pm 6$	$20 \pm 3$	12 ± 4	0.089
Isobutyl acetate (µg/L)	$81 \pm 50$	$60 \pm 16$	$94 \pm 34$	$57 \pm 21$	$84 \pm 32$	$64 \pm 6$	$85 \pm 37$	0.812
Isoamyl acetate (µg/L)	2630 ± 1859	$2009 \pm 726$	1919 ± 209	$1556 \pm 778$	$3690 \pm 2796$	1214 ± 618	2704 ± 1620	0.511
Hexyl acetate (µg/L)	523 ± 2581	$305 \pm 167$	$262 \pm 90$	$250 \pm 84$	301 ± 285	158 ± 97	$325 \pm 130$	0.102
Diethyl succinate (µg/L)	1142 ± 835	$1299 \pm 359$	2201 ± 953	1802 ± 832	1441 ± 551	$1979 \pm 669$	1241 ± 766	0.351
Alcohols								
1-Hexanol (µg/L)	$2533 \pm 603$	2506 ± 634	2506 ± 202	$2363 \pm 738$	1524 ± 662	1723 ± 271	2100 ± 689	0.147
3-Hexenol (E) (µg/L)	92 ± 20	$72 \pm 36$	$85 \pm 36$	$114 \pm 49$	$91 \pm 39$	70 ± 241	$47 \pm 34$	0.192
3-Hexenol (Z) (µg/L)	$521 \pm 95$ h,w,a,c,f,u	356 ± 185 °	266 ± 109	$383 \pm 78  c,u$	165 ± 47 f	$332 \pm 31$	$202 \pm 27$	< 0.0001
Isobutanol (mg/L)	$15 \pm 1^{w,f}$	16 ± 1 <sup>f</sup>	21 ± 8	18 ± 4 <sup>f</sup>	19 ± 2.6 f	27 ± 11 <sup>u</sup>	16 ± 2	0.004
Isoamylalcohol (mg/L)	201 ± 171 w,c,f	219 ± 18	$238 \pm 63$	$229 \pm 20$	$259 \pm 5.8^{\mathrm{u}}$	251 ± 13	211 ± 6.2	0.009
Phenylethanol (mg/L)	23 ± 11	23 ± 10	28 ± 13	$38 \pm 36$	23 ± 2	18 ± 2	18 ± 2	0.659
Methionol (µg/L)	$534 \pm 94$	$529 \pm 87$	$607 \pm 242$	$567 \pm 87$	$728 \pm 245$	$563 \pm 55$	632 ± 185	0.482
Thiols								
3MHA (ng/L)	395 ± 631 h,w,a,f	$74 \pm 83$	$35 \pm 28$	$37 \pm 19$	146 ± 124 f	10 ± 11	246 ± 393	0.007
3MH (ng/L)	3786 ± 1691 a,f,u	1532 ± 929	1184 ± 334	524 ± 381	2335 ± 2111	1151 ± 776	2033 ± 2874	0.013
4MMP (ng/L)	19 ± 13	$13 \pm 4.6$	24 ± 13	$8 \pm 4$	8 ± 2	11 ± 10	20 ± 19	0.143
Methoxypyrazines								
MIBP (ng/L)	$27.2 \pm 9.3$	21.0 ± -	19.5 ± 3	27.6 ± -	16.4 ± 2.8	_	15.8 ± 1.6	0.809

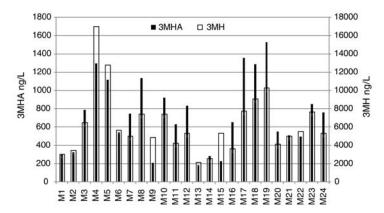
<sup>\*</sup>Superscript letters (m = Marlborough, h = Hawkes Bay, w = Wairarapa, a = Australia, c = Chile, f = France, u = United States) indicates the significant differences between countries and regions obtained using a pair-wise comparison (Tukey's HSD) with 95% level of confidence.

States wines were close together and separated from the other groups given their higher concentrations of 3-hexenol (*E*) and lower levels of most other compounds. France was well separated from all of the other groups, most likely due to lower concentrations of 3MHA, 3-hexenol (*Z*), and hexyl acetate. The third dimension explained a further 20% of the difference in the data set (Figure 3B) and clearly separated France, United States, and South Africa wines from Australia wines, which had higher concentrations of 3-hexenol (*E*), and MIPP. The 95% confidence intervals of the means for the Marlbor-





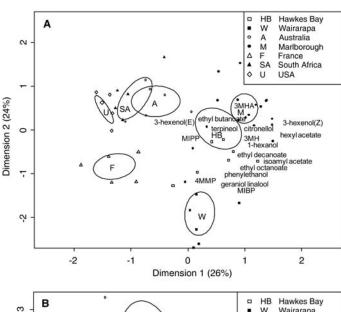
**Figure 1** 3-Mercaptohexanol (3MH) and 3-mercaptohexyl acetate (3MHA) in the Sauvignon blanc wines from the first set of samples: minima, lower quartile, median, upper quartile, and maxima.

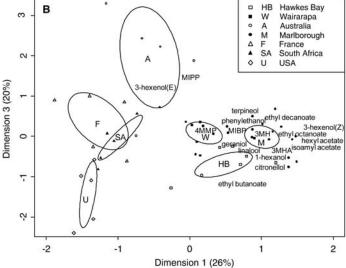


**Figure 2** 3-Mercaptohexanol (3MH) and 3-mercaptohexyl acetate (3MHA) concentrations in the commercial Marlborough Sauvignon blanc wines from the 2007 vintage.

ough and Hawkes Bay regions were well separated due to 3MH, 3MHA, ethyl octanoate, ethyl decanoate, 1-hexanol, citronellol, 3-hexenol (*Z*), and hexyl acetate.

Because the CVA technique assumes a normal distribution of the variables, which cannot be expected for our very diverse set of samples, the data set was also submitted to a classification tree analysis. This analysis obtained for the same data set (Figure 4) again discriminated the New Zealand wines from the international wines. The most powerful variable was 3MHA, which separated the Marlborough and Wairarapa wines from all other wines. On the next level, 3-hexenol (*E*) discriminated between the Marlborough and Wairarapa wines. Phenylethanol discriminated samples from the United States, South Africa, and Australia from the New Zealand and French wines. The French wines were well separated from the New Zealand wines, except for a few Wairarapa wines, due to lower concentrations of 3-hexenol (*Z*) and



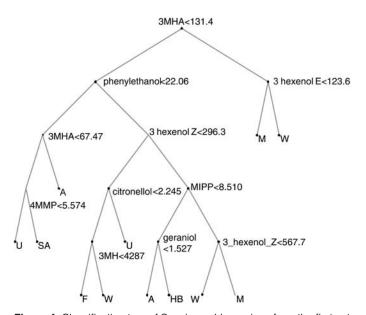


**Figure 3** Regional differentiation of Sauvignon blanc wines from the first set of samples (2003 and 2004 vintages) by canonical variate analysis. Score and loadings plots of (A): first and second dimensions; (B): first and third dimensions.

3MH. Some samples from Australia were classified closer to the New Zealand wines than to the other international wines, with MIPP and geraniol as the variables.

These results raise the issue of what impact the differences in the chemical composition among regions and countries may have on the sensory perception of these wines. Quantitative descriptive analysis of 16 descriptors was available for the first set of samples (2003 and 2004 vintage), as reported previously (Lund et al. 2009a). A partial least squares regression (PLSR) was presented using the chemical data for three volatile compounds (3MH, 3MHA, and MIBP) (X variables) and the sensory data (Y variables) to determine correlations between chemical concentrations and sensory properties of the wines (Lund et al. 2009a). A second PLSR analysis of the previously published sensory data (Lund et al. 2009a) used a higher number of volatile compounds as X variables (Figure 5); 3MHA and 3MH showed correlations within the 95% confidence ellipse with their specific sensory descriptors: tropical, sweet sweaty passion fruit, and passion fruit skin stalk (Figure 5A). There were additional correlations among the chemical data between the two ethyl esters, ethyl octanoate and ethyl decanoate, and between the two acetates, hexyl acetate and isoamyl acetate. However, no other positive correlations between sensory and chemical data were observed within the 95% confidence ellipse.

Additional information concerning the impact of the chemical composition on the sensory perception of the wines was obtained from the score and X loadings (wines and chemical data) plot of the PLSR analysis (Figure 5B). For the CVA analysis, the discrimination between the New Zealand wines (right side of the plot, with a few exceptions) and the international wines (left side of the plot) is obvious and is due to higher concentrations of the chemicals quantified, except 3-hexenol (*E*) in most of the New Zealand wines. The descriptors tropical, sweet sweaty passion fruit, pas-

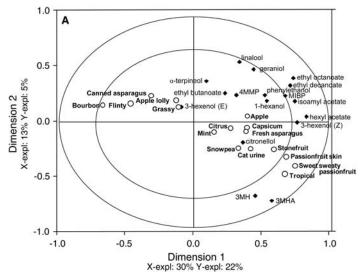


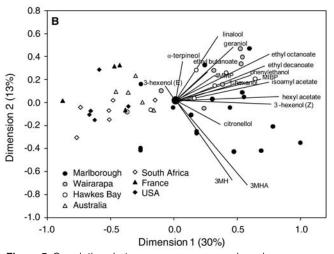
**Figure 4** Classification tree of Sauvignon blanc wines from the first set of samples (2003 and 2004 vintages).

sion fruit skin/stalk, stone fruit, cat urine, snow pea, fresh asparagus, citrus, mint, capsicum, and apple are grouped on the right side of the plot (Figure 5A) in the direction of the New Zealand wines, and higher concentrations of most of the aroma chemicals assigned to these characters were found in the New Zealand wines (Figure 5B). On the other hand, descriptors such as bourbon, flinty, canned asparagus, apple lolly, and grassy were more intense in the international wines and were grouped on the left side of the plot. Finally, the second dimension of the PLSR plot (Figure 5A, B) effectively discriminated a number of the Marlborough wines in the lower left quarter of the plot based on higher concentrations of 3MHA and 3MHA and higher intensities of the descriptors tropical, sweet sweaty passion fruit, and passion fruit skin stalk, together with other descriptors such as stone fruit, cat urine, snow pea, fresh asparagus, citrus, mint, and capsicum.

## **Discussion**

Based on perception thresholds and odor activity values, past research has largely attributed the characteristic aroma of





**Figure 5** Correlations between aroma compounds and sensory properties of the Sauvignon blanc wines from the first set of samples (2003 and 2004 vintages) by partial least squares regression analysis. (**A**): first and second dimensions of the loadings (X = chemical data; Y = sensory data); (**B**): score and X loadings (wines/chemical data).

Sauvignon blanc wines to a few aroma compounds, represented by the polyfunctional mercaptans and methoxypyrazines (Tominaga et al. 1998a, 2006, Augustyn et al. 1982, Allen et al. 1991, Lacey et al. 1991, Marais 1994). However, the use of perception thresholds is questionable because of the use of synthetic matrices instead of wine matrices for threshold determinations. Moreover, recent research approaches, such as aroma reconstitution (Escudero et al. 2004, 2007, Campo et al. 2005, Ferreira et al. 2006, Pineau et al. 2007), indicate that wine aroma is influenced by complex interactions between various wine components and is rarely dominated by a single component. In order to understand what chemicals contribute to the distinct aroma characteristics of Marlborough Sauvignon blanc wine, this study has taken a holistic approach to Sauvignon blanc wine aroma by quantifying an important number of volatile compounds in wines over two vintages. Among the 20 volatile compounds quantified (Table 4), several compounds other than the polyfunctional mercaptans and methoxypyrazines were shown to be present in concentrations above their sensory perception thresholds (Table 2). In particular, several fermentative esters—isoamyl acetate, ethyl butanoate, ethyl hexanoate, ethyl isovalerate, and ethyl isobutyrate—exceeded perception thresholds. Other esters such as hexyl acetate, ethyl octanoate, and ethyl decanoate were very close to their reported perception thresholds. With fruity, flowery, and sweet characters, esters are known to be particularly important in the generation of the bouquet of young white wines and interact with an additive effect (van der Merwe and van Wyk 1981, Campo et al. 2005). Moreover, the esters contributed to the separation of New Zealand wines from the international wines (Figure 3, Figure 5) in the first, but not the second, set of samples. More importantly, there was a positive relationship between concentrations of higher esters and a higher intensity of the fruity, capsicum, and fresh asparagus descriptors and a negative relationship with a higher intensity of the descriptors bourbon, flinty, and canned asparagus (Figure 5). The green, grassy-like higher alcohols, 1-hexanol and 3-hexenol (Z), were also present at levels close to or above their perception thresholds (Table 2). These two compounds also contributed to the discrimination between most of the New Zealand and the international wines (Figure 3). Moreover, 3-hexenol (Z) was also involved in separating the New Zealand and some Australia wines from the other wines in the classification tree analysis (Figure 4). Despite quantification at concentrations close to or just under their perception thresholds (Table 2), the higher alcohols and monoterpenes contributed to the regional discrimination of the wines (Figure 3, Figure 5), with phenylethanol as the second most powerful variable that separated the wines using the classification tree analysis. The varietal methoxypyrazines, especially MIBP, were present at concentrations well above their perception thresholds, and they contributed to the regional discrimination of the first set of wines (vintage 2003 and 2004). However, as reported previously (Lund et al. 2009a), MIBP did not exhibit a direct correlation with the capsicum descriptor (Figure 5), but was involved in a negative relationship with the intensity rating of the bourbon character.

A masking or suppressing effect between alkyl methoxypyrazines and 3MHA (Campo et al. 2005) and white wine polyphenols (Lund et al. 2009b) has been reported already and could also explain these results.

Finally, for all wines, the varietal polyfunctional mercaptans were present at concentrations well above their perception thresholds (Table 2). Moreover, 3MH (passion fruit skin/ stalk) and 3MHA (sweet sweaty passion fruit) were typically present at higher concentrations in the wines from the Marlborough region, together with some wines from Wairarapa (first set of samples), United States, and Chile (second set of samples). The importance of 3MHA and 3MH for the discrimination of Marlborough wines was confirmed by the classification tree (Figure 4) and CVA (Figure 3) approaches. The Marlborough wines also showed a high degree of variability in 3MH and 3MHA concentrations (Figure 1, Figure 2), which were the only compounds that correlated well with their sensory descriptors, as demonstrated by PLSR analysis (Figure 5). Compared to 3MH and 3MHA, the boxwood/cat urine-like 4MMP displayed a different trend, with a much lower variation within and between regions. Likewise, 4MMP played a less important role in discriminating the Marlborough wines. A strong correlation could not be obtained between 4MMP and the cat urine or any other sensory descriptor.

Wine sensory profiles and/or wine chemical composition have been successfully and extensively used to investigate geographical influences on wines made from several grape varieties, such as Riesling (Douglas et al. 2001), Chardonnay (Schlosser et al. 2005), Albarino (Vilanova and Vilarino 2006), Pinot noir (Cliff and Dever 1996), and Sauvignon blanc (Berna et al. 2009). This study is the first to consider and successfully demonstrate the implication of a comprehensive number of wine volatile compounds in the geographic discrimination of Sauvignon blanc wines. Our results confirmed the clear role for 3MH and 3MHA in the aroma of Sauvignon blanc wines and in the discrimination of wines. However, both CVA (Figure 3) and PLSR analyses (Figure 5) indicate the contribution of several other compounds to Sauvignon blanc aroma. The discrimination of New Zealand wines from the international wines was based on a number of compounds that also included methoxypyrazines, 4MMP, fermentative esters, C<sub>6</sub>-alcohols, phenylethanol, and monoterpenes. While the regional differences obtained in this study will be influenced by the inevitable bottling time differences between the two hemispheres, differences in winemaking practices, and an unbalanced sampling in favor of the Marlborough region, the overall trend of wine chemical and sensory data separation is still apparent and is strongly supported by the PLSR analysis (Figure 5). This analysis separated the wines by differences in the concentrations of volatile compounds and in the complex sensory profiles, with higher intensities of the tropical, sweet sweaty passion fruit, passion fruit skin/stalk, stone fruit, cat urine, snow pea, fresh asparagus, capsicum, apple, citrus, and mint in the Marlborough wines. On the other hand, the wines that had lower concentrations of most of the volatile compounds were also characterized by a different sensory profile, with lower

intensities of the descriptors listed above, and were instead dominated by bourbon, flinty, canned asparagus, apple lolly, and grassy-like descriptors. Notably, the wines that were well separated due to higher intensities of the tropical, sweet sweaty passion fruit, and passion fruit skin/stalk descriptors were all Marlborough wines that had exceptionally high concentrations of 3MH (>5000 ng/L) and 3MHA (>500 ng/L). Additional Marlborough wine profiles were evident, but they were more difficult to define and seem to be influenced by more volatile compounds. This result is not surprising, as Marlborough Sauvignon blanc wines are known to exhibit different styles ranging from a grassy, crispy end to a more tropical, fruity style. This important intraregional variation can be associated with great differences in 3MH and 3MHA concentrations (Figure 1, Figure 2). Further research on the composition of Marlborough Sauvignon blanc using wines obtained under a controlled manner (similar grape pressing conditions and standardized winemaking) may help to understand this variation in 3MH and 3MHA concentrations.

Several studies have used PLSR to demonstrate correlations between wine volatile compounds and sensory characteristics (Noble and Ebeler 2002, Aznar et al. 2003). The PLRS analysis was helpful for this study in establishing several relationships between the sensory and chemical data together with the importance of 3MH and 3MHA to the aroma of certain Marlborough Sauvignon blanc wines. However, the correlations obtained were too weak to be used for Sauvignon blanc aroma predictive models, despite intensive training of the sensory panelists with chemical references and specifically selected descriptors (Lund et al. 2009a). Therefore, this study demonstrates the limitation of this approach when the chemical compounds are considered for their contribution to wine aroma quality. Existing statistical tools that allow the correlation of chemical and sensory data to be examined, such as PLRS, are based on linear regression models and therefore cannot take into consideration a nonlinear relationship between sensory descriptors and chemical compounds (Guichard 1995, Meilgaard 2006), changes of the sensory descriptor or sensory descriptor intensity related to the chemical concentration of a specific compound, and/or any interaction with other compounds from the wine matrix (Murat et al. 2001, Campo et al. 2005, Escudero et al. 2007, Pineau et al. 2007, Lund et al. 2009b). From the point of view of the results obtained in the present study, there is a clear need for a deeper understanding of Sauvignon blanc wine aroma complexity through methodologies such as aroma reconstitution, designed to confirm the specific role of individual compounds in different wine matrices. These research methodologies can be used the confirm the role of different concentrations of 3MH and 3MHA in Sauvignon blanc wine profiles and to understand the contribution of C<sub>6</sub>-alcohols, fermentative esters, methoxypyrazines, phenylethanol, and monoterpenes to the overall aroma of Sauvignon blanc wines.

## **Conclusions**

The quantification of numerous volatile compounds in Sauvignon blanc wines from New Zealand and several in-

ternational producers over several vintages showed the importance of polyfunctional mercaptans, methoxypyrazines, and fermentative esters and C<sub>6</sub>-alcohols for wine aroma. Multivariate statistical analysis was able to discriminate New Zealand wines from most of the international wines, based on the concentrations of polyfunctional mercaptans, C<sub>6</sub>-alcohols, fermentative esters, methoxypyrazines, phenylethanol, and some monoterpenes. When the sensory data was considered through a PLS analysis, higher amounts of these volatile compounds could explain the more complex aroma profiles dominated by higher intensities of the tropical, sweet sweaty passion fruit, passion fruit skin/stalk, stone fruit, cat urine, snow pea, fresh asparagus, capsicum, apple, citrus, and mint descriptors. The role of the polyfunctional mercaptans, 3MHA and 3MH, in the distinctive tropical, fruity character of Marlborough Sauvignon blanc wine was strongly demonstrated. Moreover, a high concentration of these compounds was also characteristic of the Marlborough wines, although here again the contribution of the other volatile compounds needs to be considered. A more complete understanding of Sauvignon blanc chemical composition and aroma reconstitution is the focus of our future research to elucidate the contribution of the different chemicals involved in the overall aroma profiles of this distinctive wine.

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